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## Key indicators

Single-crystal X-ray study  
 $T = 298\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$   
 $R$  factor = 0.039  
 $wR$  factor = 0.102  
Data-to-parameter ratio = 9.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(+)-*N*-[(*R*)-1-(2-Hydroxy-5-methylphenyl)propyl]-*N*-  
[(*R*)-2-methyl-1-phenylpropyl]ammonium chloride**

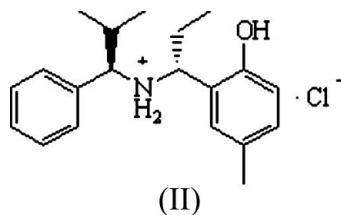
The title compound,  $\text{C}_{20}\text{H}_{28}\text{NO}^+\cdot\text{Cl}^-$ , was synthesized by a condensation reaction. The absolute configuration of the new stereogenic centre (the C atom between the N atom and the phenol ring) was determined as *R*. The crystal structure is stabilized through  $\text{N}-\text{H}\cdots\text{Cl}$  and  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonds and intramolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonding.

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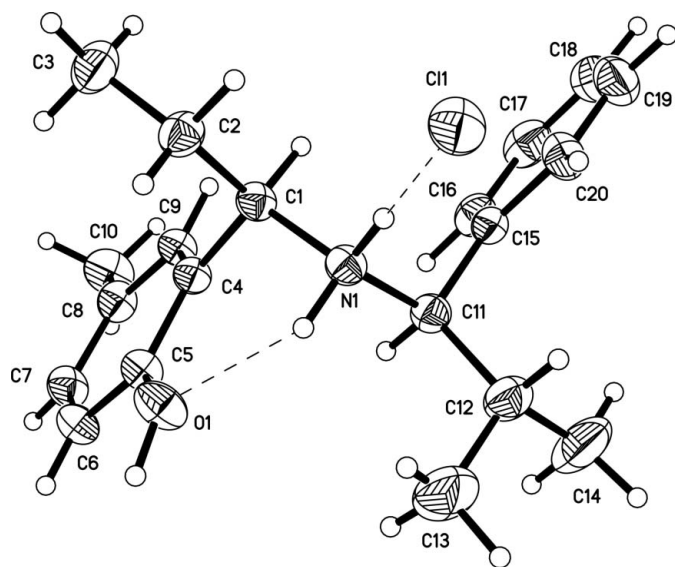
## Comment

The search for new chiral ligands to be used in asymmetric catalysis is of great interest in the field of synthetic chemistry (Pu & Yu 2001; Cimarelli *et al.*, 2002). Carbon-carbon bond-forming reactions remain an active area of research (Le Goanvic *et al.*, 2002). A popular reaction is that of dialkylzinc with pro-chiral aldehydes (Le Goanvic *et al.*, 2002; Vilaplana *et al.*, 2002; Superchi *et al.*, 2002), since the chiral secondary alcohols that are formed are important substrates for drug synthesis. Amino alcohols are excellent ligands for this reaction, as shown by Noyori & Kitamura (1991). 1,3-Aminophenols are analogous to amino alcohols, although they were not used as chiral ligands in asymmetric catalysis until 1998 (Cardellicchio *et al.*, 1998, 1999), when they showed high enantioselectivity. Since then, interest in the synthesis of aminophenols and their derivatives, and in their application in asymmetric catalysis, has increased significantly (Lu *et al.*, 2002; Li *et al.*, 2003; Zhang *et al.*, 2003; Yang, Zhang, Zhang *et al.*, 2005). Following on from our previous work (Yang, Zhang, Liu *et al.*, 2005), we prepared a new aminoalkylphenol, namely 4-methyl-2-[(*R*)-1-((1*R*)-2-methyl-1-phenylpropylamino)-propyl]phenol, (I). In this paper, we report the structure of the title compound, (II), which is a salt of (I) with HCl.

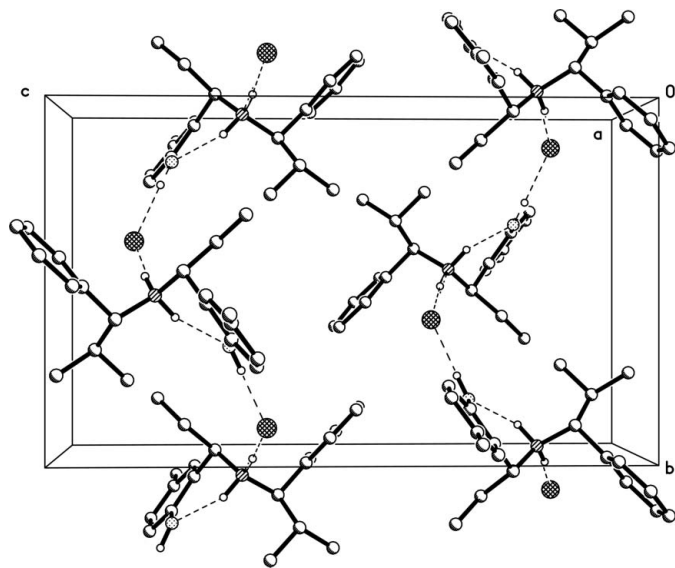


The molecular structure of (II) is shown in Fig. 1. The absolute configuration of the new stereogenic centre C1 is *R*. The chain of atoms C3/C2/C1/N1/C11/C12/C14 shows an approximately all-*trans* conformation (Table 1). The dihedral angle between the C4–C9 and C15–C20 aromatic rings is  $12.9(2)^\circ$ .

Atom N1 acts as a hydrogen-bond donor to phenol atom O1 and to the Cl1 anion (Fig. 1 and Table 2). There is also an  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bond, forming chains along the *b* axis (Fig. 2).



**Figure 1**  
The asymmetric unit of (II), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Dashed lines indicate hydrogen bonds.



**Figure 2**  
A view of the packing in (II). H atoms have been omitted, except for those involved in hydrogen bonds (dashed lines).

## Experimental

(*R*)-2-methyl-1-phenylpropylamine (3 mmol; Yang, Zhang, Liu *et al.*, 2005) and 1-(2-hydroxy-5-methylphenyl)-propan-1-one (3 mmol; purchased from J & K Chemical Ltd) were dissolved in methanol (20 ml) and reacted at room temperature for 12 h. The solvent was removed and NaBH<sub>4</sub> (15 mmol) was added to the solution in a tetrahydrofuran–ethanol (1:1 v/v, 20 ml) mixture. The reaction was allowed to stand and was monitored by thin-layer silica-gel chromatography. The reaction was stopped when complete conversion of the starting material was detected by thin-layer silica-gel chromatography. 6 M HCl was then added dropwise to the reaction mixture until hydrogen production ceased and the mixture was then neutralized with Na<sub>2</sub>CO<sub>3</sub>. The aqueous solution was extracted with CHCl<sub>3</sub>. The organic layer was dried with anhydrous sodium sulfate and filtered. The solvent was removed under reduced pressure. Further purification was carried out by thin-layer silica-gel chromatography (first eluted with chloroform–methanol, 60:1, v/v; second eluted with chloroform) to give the chiral compound, (I) {canary-yellow oil, [α]<sub>D</sub><sup>25</sup> = 52.2° (c = 0.5, CHCl<sub>3</sub>)}. The chemical yield of (*R,R*)-(I) and the diastereoisomeric excess {measured by chiral high-performance liquid chromatography [Chiralcel OD-H, propan-2-ol–hexane (5:95, v/v) of the corresponding amide derivatives of racemic and chiral (I)]} are 76.2% and 96.0%, respectively.

Compound (I) and concentrated HCl were reacted at room temperature and a colourless solid was precipitated. The solvent was removed and the solid residue was recrystallized from an ethanol–hexane (7:3, v/v) mixture to yield compound (II) (m.p. 461–462 K).

## Crystal data

C <sub>20</sub> H <sub>28</sub> NO <sup>+</sup> ·Cl <sup>-</sup>	Z = 4
M <sub>r</sub> = 333.88	D <sub>x</sub> = 1.131 Mg m <sup>-3</sup>
Orthorhombic, P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	Mo Kα radiation
a = 8.725 (2) Å	μ = 0.20 mm <sup>-1</sup>
b = 11.641 (3) Å	T = 298 (2) K
c = 19.313 (5) Å	Block, colourless
V = 1961.7 (9) Å <sup>3</sup>	0.33 × 0.21 × 0.13 mm

## Data collection

Bruker SMART CCD area-detector diffractometer	10299 measured reflections
φ and ω scans	1975 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1503 reflections with I > 2σ(I)
T <sub>min</sub> = 0.937, T <sub>max</sub> = 0.975	R <sub>int</sub> = 0.038
	θ <sub>max</sub> = 25.0°

## Refinement

Refinement on F <sup>2</sup>	w = 1/[σ <sup>2</sup> (F <sub>o</sub> <sup>2</sup> ) + (0.0453P) <sup>2</sup> + 0.4485P]
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )] = 0.039	where P = (F <sub>o</sub> <sup>2</sup> + 2F <sub>c</sub> <sup>2</sup> )/3
wR(F <sup>2</sup> ) = 0.102	(Δ/σ) <sub>max</sub> < 0.001
S = 1.05	Δρ <sub>max</sub> = 0.20 e Å <sup>-3</sup>
1975 reflections	Δρ <sub>min</sub> = -0.16 e Å <sup>-3</sup>
208 parameters	Absolute structure: Flack (1983), with 1469 Friedel pairs
H-atom parameters constrained	Flack parameter: -0.07 (9)

**Table 1**

Selected torsion angles (°).

C11–N1–C1–C2	-169.4 (3)	C1–N1–C11–C12	-168.3 (3)
N1–C1–C2–C3	177.8 (3)	N1–C11–C12–C14	176.3 (4)

**Table 2**

Hydrogen-bond geometry (Å, °).

D–H...A	D–H	H...A	D...A	D–H...A
N1–H1A...Cl1 <sup>i</sup>	0.90	2.27	3.144 (3)	163
N1–H1B...O1	0.90	2.06	2.737 (4)	131
O1–H1...Cl1 <sup>ii</sup>	0.82	2.24	3.032 (3)	161

Symmetry codes: (i) x + 1, y, z; (ii) -x + 1, y + 1/2, -z + 3/2.

All H atoms were placed in idealized positions and constrained to ride on their parent atoms, with C–H = 0.93–0.98 Å, N–H = 0.90 Å and O–H = 0.82 Å, and with U<sub>iso</sub>(H) = 1.2U<sub>eq</sub>(C,N), or 1.5U<sub>eq</sub>(C,O) for methyl and hydroxy groups.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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## References

- Bruker (1997). *SMART* (Version 5.059), *SAINTE* (Version 6.01) and *SHELXTL* (Version 5.10). Bruker AXS Inc., Madison, Wisconsin, USA.
- Cardellicchio, C., Ciccarella, G., Naso, F., Perna, F. & Tortorella, P. (1999). *Tetrahedron*, **55**, 14685–14692.
- Cardellicchio, C., Ciccarella, G., Naso, F., Schingaro, E. & Scordari, F. (1998). *Tetrahedron Asymmetry*, **9**, 3667–3675.
- Cimarelli, C., Palmieri, G. & Volpini, E. (2002). *Tetrahedron Asymmetry*, **13**, 2417–2426.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Le Goanvic, D., Holler, M. & Pale, P. (2002). *Tetrahedron Asymmetry*, **13**, 119–121.
- Li, X.-S., Sheng, Z.-Q., Ge, J.-F. & Zhang, Y.-W. (2003). *Chin. J. Org. Chem.* **23**, 387–389. (In Chinese).
- Lu, J., Xu, X.-N., Wang, S.-Z., Wang, C., Hu, Y.-F. & Hu, H.-W. (2002). *J. Chem. Soc. Perkin Trans. 1*, pp. 2900–2903.
- Noyori, R. & Kitamura, M. (1991). *Angew. Chem. Int. Ed. Engl.* **30**, 49–69.
- Pu, L. & Yu, H. L. (2001). *Chem. Rev.* **101**, 757–824.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Superchi, S., Giorgio, E., Scafato, P. & Rosini, C. (2002). *Tetrahedron Asymmetry*, **13**, 1385–1391.
- Vilaplana, M. J., Molina, P., Arques, A., Andres, C. & Pedrosa, R. (2002). *Tetrahedron Asymmetry*, **13**, 5–8.
- Yang, X.-F., Zhang, G.-Y., Liu, C.-X. & Zhang, Y. (2005). *Chem. Res. Appl.* **17**, 249–251. (In Chinese).
- Yang, X.-F., Zhang, G.-Y., Zhang, Y., Zhao, J.-Y. & Wang, X.-B. (2005). *Acta Cryst.* **C61**, o262–o264.
- Zhang, G.-Y., Liao, Y.-Q., Wang, Z.-H., Nohira, H. & Hirose, T. (2003). *Tetrahedron Asymmetry*, **14**, 3297–3300.